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Corrigendum

Corrigendum to "Oxidation of adamantane with 1 atm molecular oxygen by vanadium-substituted polyoxometalates" [J. Catal. 233 (1) (2005) 81–89]

Satoshi Shinachi, Mitsunori Matsushita, Kazuya Yamaguchi, Noritaka Mizuno *

Department of Applied Chemistry, School of Engineering, The University of Tokyo, 7-3-1 Hongo, Bunkyo-ku, Tokyo 113-8656, Japan

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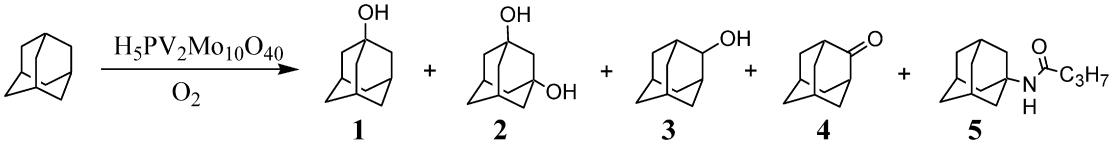
The authors regret that Tables 1 and 2 in the article cited above included some incorrect entries and should be replaced with Tables 1 and 2 below.

Table 1
Oxidation of adamantane with molecular oxygen in various solvents[a]



1 **2** **3** **4** **5**

Entry	Solvent	Yield (%)	Selectivity (%)				
			1	**2**	**3**	**4**	**5**
1	Butyronitrile	46	54	7	4	16	19
2[b]	Butyronitrile	84	43	24	1	20	12
3	Diethylketone	30	65	11	7	17	–
4	Cyclopentanone	22	68	4	12	16	–
5	Dimethylformamide	12	80	–	9	11	–
6	Acetic acid	11	65	–	15	20	–
7	Acetonitrile	<1	–	–	–	–	–
8	Toluene	<1	–	–	–	–	–
9	1,2-Dichloroethane	<1	–	–	–	–	–

[a] Reaction conditions: $H_5PV_2Mo_{10}O_{40}$ (2 μmol), adamantane (1 mmol), solvent (3 mL), 356 K, 96 h under 1 atm of molecular oxygen. Yields and selectivities were determined by gas chromatographic analysis using naphthalene as an internal standard.
[b] 288 h.

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* Corresponding author. Fax: +81 3 5841 7220.
E-mail address: tmizuno@mail.ecc.u-tokyo.ac.jp (N. Mizuno).

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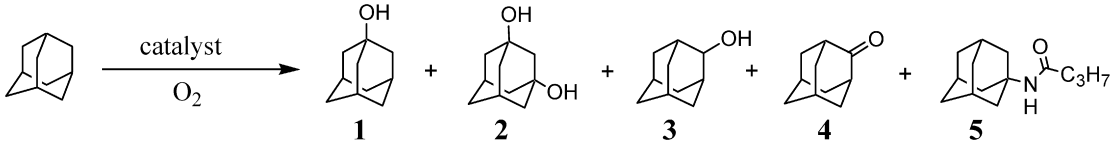
Table 2
Oxidation of adamantane with molecular oxygen in the presence of various catalysts[a]



Entry	Catalyst	Yield (%)	Selectivity (%)					C^3–H/C^2–H[b]
			1	**2**	**3**	**4**	**5**	
1	$H_5PV_2Mo_{10}O_{40}$	46	54	7	4	16	19	13.1
2	$H_6PV_3Mo_9O_{40}$	46	54	7	4	16	19	13.1
3	$H_4PVMo_{11}O_{40}$	39	54	6	5	15	20	12.9
4	$H_4PVW_{11}O_{40}$	26	62	5	6	15	12	12.0
5	$H_3PMo_{12}O_{40}$	7	50	8	8	12	22	13.2
6	$H_3PW_{12}O_{40}$	<1	–	–	–	–	–	–
8	$VO(acac)_2$	29	78	–	–	22	–	10.6
9	$Co(OAc)_2$	17	80	2	7	11	–	14.0
10	$Mn(OAc)_2$	<1	–	–	–	–	–	–
11	None	<1	–	–	–	–	–	–

[a] Reaction conditions: catalyst (2 μmol), adamantane (1 mmol), butyronitrile (3 mL), 356 K, 96 h under 1 atm of molecular oxygen. Yields and selectivities were determined by gas chromatographic analysis using naphthalene as an internal standard. Carbon balance for each reaction was more than 93%.

[b] The selectivity parameter defined by the relative reactivity of tertiary C–H bonds to secondary C–H bonds ($= \{(\mathbf{1} + 2 \times \mathbf{2} + \mathbf{5})/4\}/\{(\mathbf{3} + \mathbf{4})/12\}$).